

## Subject Description Form

<b>Subject Code</b>	ME34001
<b>Subject Title</b>	Engineering Thermodynamics
<b>Credit Value</b>	3
<b>Level</b>	3
<b>Pre-requisite/ Co-requisite/ Exclusion</b>	Pre-requisite: Nil
<b>Objectives</b>	To provide fundamental knowledge of steam and gas power cycles and refrigeration cycle, and air-conditioning, combustion and heat transfer processes.
<b>Intended Learning Outcomes</b>	<ol style="list-style-type: none"> <li>a. Formulate and solve thermodynamic problems relating to steam power, gas power and refrigeration cycles; and air-conditioning, combustion and heat-transfer processes by applying knowledge in engineering thermodynamics, air-conditioning, combustion, heat-transfer and mathematics.</li> <li>b. Complete a given task such as a design project in thermodynamics by applying knowledge acquired in the subject and information obtained through literature search.</li> <li>c. Analyze and interpret data obtained from experiments in engineering thermodynamics, combustion and heat transfer.</li> <li>d. Present effectively in completing written reports of laboratory work and the given task.</li> </ol>
<b>Subject Synopsis/ Indicative Syllabus</b>	<p><b>Review of Basic Concepts of Thermodynamics</b> - Thermal properties. Ideal gas. First law of thermodynamics. Non-flow and steady-flow processes. Second law of thermodynamics.</p> <p><b>Second Law of Thermodynamics</b> - Kelvin-Planck and Clausius statements. Reversible and irreversible processes. Carnot cycle. Thermodynamic temperature scale. Inequality of Clausius. Entropy. The second law for a control mass/control volume. Isentropic efficiency.</p> <p><b>Power and Refrigeration Cycles</b> - Vapour cycles: Carnot cycle and Rankine cycle. Superheat and reheat. Air standard engine cycles: Otto cycle and Diesel cycle. Gas turbine cycles. Carnot efficiency. Refrigerator and heat pump. Vapor compression cycle. Coefficient of Performance.</p> <p><b>Psychrometry and Air Conditioning</b> - Psychrometry. Psychrometric chart. Introduction to air conditioning.</p> <p><b>Combustion</b> - Hydrocarbon fuels. Combustion equations. Stoichiometric air fuel ratio. Lean and rich mixture.</p> <p><b>Review of Fundamental Heat Transfer</b> - Mechanisms and governing equations of conduction, convection and radiation.</p> <p><b>Convection Heat Transfer</b> - Forced, free and mixed convection. Hydrodynamic and thermal boundary layers. Use of non-dimensional parameters: Reynolds number;</p>

	<p>Nusselt number; Prandtl number; Grashof number and Richardson number. Application of convective heat transfer correlations to solve convective heat transfer problems.</p> <p><b>Laboratory Experiment</b>  There are two 2-hour laboratory sessions with the typical experiments:</p> <ol style="list-style-type: none"> <li>1. Refrigeration system</li> <li>2. Diesel engine test-bed</li> <li>3. Convection heat transfer</li> <li>4. Combustion</li> </ol>																																								
<p><b>Teaching/Learning Methodology</b></p>	<p>Lectures are used to deliver the fundamental knowledge in relation to thermodynamics and heat transfer (outcomes a and b).</p> <p>Tutorials are used to illustrate the application of fundamental knowledge to practical situations (outcomes a, b and d).</p> <p>Experiments are used to relate the concepts to practical applications and students are exposed to hand-on experience, proper use of equipment and application of analytical skills on interpreting experimental results (outcomes c and d).</p> <table border="1" data-bbox="443 880 1442 1140"> <thead> <tr> <th rowspan="2">Teaching/Learning Methodology</th> <th colspan="4">Outcomes</th> </tr> <tr> <th>a</th> <th>b</th> <th>c</th> <th>d</th> </tr> </thead> <tbody> <tr> <td>Lecture</td> <td>√</td> <td>√</td> <td></td> <td></td> </tr> <tr> <td>Tutorial</td> <td>√</td> <td>√</td> <td></td> <td>√</td> </tr> <tr> <td>Experiment</td> <td></td> <td></td> <td>√</td> <td>√</td> </tr> </tbody> </table>	Teaching/Learning Methodology	Outcomes				a	b	c	d	Lecture	√	√			Tutorial	√	√		√	Experiment			√	√																
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<b>Student Study Effort Expected</b>	Class contact:	
	▪ Lecture	33 Hrs.
	▪ Tutorial / Experiment	6 Hrs.
	Other student study effort:	
	▪ Course work	40 Hrs.
	▪ Self-study	38 Hrs.
	Total student study effort	117 Hrs.
<b>Reading List and References</b>	<ol style="list-style-type: none"> <li>1. R.E. Sonntag, C. Borgnakke and G.J.V. Wylen, Fundamentals of Thermodynamics, John Wiley and Son, latest edition.</li> <li>2. T.D. Eastop and A. McConkey, Applied Thermodynamics for Engineering Technologists, Pearson, latest edition.</li> <li>3. K. Wark, and D. Richards, Thermodynamics, McGraw-Hill, latest edition.</li> <li>4. K.D. Hagen, Heat Transfer with Applications, Prentice Hall, latest edition.</li> <li>5. F.D. Incropera, and D.P. Dewitt, Introduction to Heat Transfer, Wiley, latest edition.</li> </ol>	

*Revised July 2018*