

## Subject Description Form

<b>Subject Code</b>	ABCT4777
<b>Subject Title</b>	CHEMICAL & BIOPROCESS TECHNOLOGY
<b>Credit Value</b>	3
<b>Level</b>	4
<b>Pre-requisite</b>	INTRODUCTION TO CHEMICAL & BIOPROCESS TECHNOLOGY
<b>Objectives</b>	To introduce the general chemical & bioprocess engineering principles, with a focus on the common processes and operations in the chemical and bioprocess industry.
<b>Intended Learning Outcomes</b>	<p>Upon completion of the subject, students will be able to:</p> <ol style="list-style-type: none"> <li>understand the theoretical principles and practical considerations for design and operation of chemical and biological processes, and the engineering approaches to deriving the design equations for complex processes.</li> <li>design and predict the major process parameters in chemical, biochemical and separation processes.</li> <li>use and analyze experimental data to derive the kinetic and process parameters with simple computing techniques.</li> </ol>
<b>Subject Synopsis/ Indicative Syllabus</b>	<p><u>Chemical Reactors and Bioreactors</u> Batch reactor and continuous stirred tank reactor (CSTR); fixed and fluidized beds. Batch growth (fermentation) kinetics; continuous bioreactor (chemostat); fed-batch process.</p> <p><u>Bioprocesses and bioreactors</u> Batch growth kinetics; continuous stirred-tank bioreactor (chemostat); practical considerations for choice of processes; Bioreactor mixing and mass (O<sub>2</sub>) transfer.</p> <p><u>Principles of mass transfer;</u> Mass transfer concepts: molecular diffusion and convective mass transfer, convective mass transfer coefficients; gas-liquid two phase transfer, oxygen transfer in bioreactors.</p> <p><u>Product Recovery and Purification</u> Evaporation, crystallization and drying; Mass transfer operations: extraction, gas absorption, fractional distillation.</p>
<b>Teaching/Learning Methodology</b>	<p><b>Lectures:</b> to introduce the essential contents, to elaborate the major principles, concepts and relationships and processing units. Practical examples and problems will be used to illustrate the principles.</p> <p><b>Tutorials</b> (in smaller groups): to make further explanation/clarification of the major points and difficult/problematic contents, to apply the concepts and principles in problems and exercises, and to have more interactive and effective contact and discussion with the students.</p> <p><b>After class:</b> homework assignments and exercises will be given to students.</p> <p><b>On-line resources:</b> a subject web will be set up and used as a teaching aid.</p>

	<b>Detail answers/solution manuals</b> are provided to the students for most of the assignment, test and examination questions.							
<b>Assessment Methods in Alignment with Intended Learning Outcomes</b>	Specific assessment methods/tasks		% weighting		Intended subject learning outcomes to be assessed (Please tick as appropriate)			
			a	b	c			
	1. Final exam		60	√	√	√		
	2. Course work		40	√	√	√		
	Total		100 %					
Explanation of the appropriateness of the assessment methods in assessing the intended learning outcomes: Learning outcomes will be assessed continually through written assignments, tests and exams. The connection of these assessments to the learning outcomes will be stated explicitly to the students.								
<b>Student Study Effort Expected</b>	Class contact:							
	▪ Lectures						26 Hrs.	
	▪ Tutorials						13 Hrs.	
	Other student study effort:							
	▪ Reading and revising						56 Hrs.	
	▪ Exercises & assignments						32 Hrs.	
	Total student study effort						127 Hrs.	
<b>Reading List and References</b>	<ol style="list-style-type: none"> <li>Geankoplis C J Transport Processes and Unit Operations, 4th Edition Prentice-Hall PTR 2003</li> <li>Fogler HS, Elements of Chemical Reaction Engineering, 2nd-4th ed. 1992-2006.</li> <li>Doran P: Bioprocess Engineering Principles, Academic Press, 1998</li> <li>Shuler M &amp; Kargi F: Bioprocess Engineering, Basic Concepts, Prentice Hall, 2002.</li> </ol>							